

visocolor® HE Chlorine

en

High sensitivity test kit for the determination in the range of 0.02–0.60 mg/L Cl₂

Method:

N,N-Diethyl-1,4-phenylene diamine (DPD)

Contents of test kit (*refill pack):

sufficient for 2 x 160 tests

24 g Cl ₂ -1*	1 plastic beaker for sampling
100 mL Cl ₂ -2*	2 round glass tubes with screw caps
30 mL Cl ₂ -3*	1 comparator block
1 black measuring spoon 85 mm*	1 color comparison disc Chlorine

Hazard warning:

This test does not contain any harmful substances which must be specially labelled as hazardous.

Procedure:

1. Place comparator block into the position provided in the box (see illustration).
2. Insert color comparison disc.
3. Open both round glass tubes, rinse several times with the water sample and fill up to the mark with the sample.
4. Add **1 level black measuring spoon Cl₂-1** to the right glass tube.
5. Add **12 drops Cl₂-2** to the right glass tube, close and mix.
6. Read **immediately**: Turn color disc until both colors match by transmitted light from above. Read test results from the mark on the front side of the comparator. Intermediate values can be estimated.
1st reading = free chlorine
7. Add **3 drops Cl₂-3** to the right glass tube, close and mix. Wait **2 min.** Read value as described above. **2nd reading = total chlorine**

The bound chlorine can be calculated as difference between 2nd reading (total chlorine) and 1st reading (free chlorine).

Free chlorine: dissolved, elementary chlorine, hypochlorous acid, and hypochlorite ions

Bound chlorine: inorganic and organic chloroamines

The method can be applied also for the analysis of sea water.

Disposing of the samples:

The used analysis specimens can be flushed down the drain with tap water and channelled off to the local sewage treatment works.

Interferences:

1. Higher manganese compounds simulate free chlorine.
2. Chlorine contents above 4 mg/L can destroy the red dye (low results).
3. The temperature of the water sample should be between 15 and 30 °C.
4. Rinse glass tubes several times thoroughly. Residues of Cl₂-3 can cause higher values for free chlorine!



Conversion:

0.10 mg/L Cl₂ \triangleq 0.18 mg/L ClO₂ \triangleq 0.15 mg/L OCl⁻ \triangleq 0.21 mg/L NaOCl \triangleq 0.23 mg/L Br₂ \triangleq 0.36 mg/L I₂

Note:

Determination of bromine besides chlorine: If chlorine is present in the sample, it can be destroyed by adding a spatula of glycine (approx. 20 mg) to 25 mL sample. The sample for the bromine determination is taken from this solution. Result in mg/L Cl₂ x 2.25 = mg/L Br₂.